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number of atoms. In order to calculate the mass from a given number of atoms, these steps will be reversed.

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Conversions Name Class Date Problem Solving

continued The correct conversion factor is the one that when multiplied by the given quantity causes the units to cancel. COMPUTE 45.5 g calcium chloride  
100 1 0 g mg 45 500 mg calcium chloride EVALUATE  
Are the units correct? Yes; milligrams are the desired units.

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Conversions Conversions One of the aims of chemistry is to describe changes—to tell what changed, how it changed, and what it changed into. Another aim of chemistry is to look at matter and its changes and to ask questions such as how much, how big, how hot, how many, how hard, and how long did it take.

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Stoichiometry Name Class Date Problem Solving

continued Sample Problem 1 Ammonia is made

industrially by reacting nitrogen and hydrogen under pressure, at high temperature, and in the presence of a catalyst. The equation is  $N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$ . If

4.0 mol of  $H_2$  react, how many moles of  $NH_3$  will be produced?

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problems to allow development and practice of problem-solving skills .

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Problem 1 Ammonia is made industrially by reacting  
nitrogen and hydrogen under pressure, at high  
temperature, and in the presence of a catalyst. The  
equation is  $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$ . If 4.0 mol of H

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Stoichiometry of Gases Name Class Date Problem  
Solving continued Practice 1. Complete the table  
below using the following equation, which represents  
a reaction that produces aluminum chloride.  $2\text{Al}(s) + 3\text{Cl}_2(g) \rightarrow 2\text{AlCl}_3(s)$   
Mass Al Volume Cl<sub>2</sub>  
Conditions Mass AlCl<sub>3</sub> a. excess ?

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## **liceolefilandiere**

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