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|REB| S3|Chemistry| unit 5|Lesson3:Effect of temperature, pressure and catalyst on rate of reaction| XI Chemistry - Chapter 5 || MCAT | ECAT || Sindh Board || Quick Revision || Shaheer Yousuf Khan

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5 Answers

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Mr. Christopherson - McLean County Unit District No. 5

Answers Unit 5 Chemistry Unit 5 Quest Answer Key - Weebly in excess. From here figure out the grams of
AlI₃ and you have your answer. Part C Solution: since we have moles, we calculate directly and then
convert to grams. Al and I₂ stand in a two-to-three molar relationship, so 0.009456 mol of I₂ uses
0.006304 mol of Al.

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nits have canceled to give the desired unit, and answers are expressed to the correct number of significant figures. The known pressures and the calculated pressures are about 80% of the atmospheric pressure, in the new units. Answers in Appendix E 1. Convert a pressure of 1.75 atm to kPa and to mm Hg. 2.

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5. If nobody guesses correctly, repeat Step 4 with a different clue. Repeat until a player guesses the
word or time runs out. 6. The first player to guess the word gets 1 point. If the player can use the
word in a sentence, he or she gets 1 more point. Then that player gets a turn. 7. The first player to
score 5 points wins. Guess the Word Word ...

CHAPTER RESOURCES • Chapter 2

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