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in Speed result in Change of Direction  
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SECTION 1 SHORT ANSWER

Answer the following questions in the

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space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

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chemical names and formulas to  
describe the atomic composition of  
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WRITING CHEMICAL FORMULAS

PRACTICE □ Write the formula of the  
binary ionic compound that will result  
between the following elements: □ 1.

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- zinc and sulfur
- 2. aluminum and phosphorus
- 3. strontium and chlorine
- 4. barium and oxygen
- 5. sodium and nitrogen
- 6. lithium and iodine

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Chemical Formulas and Chemical  
Compounds Molecules and Molecular  
Compounds □ A molecule consists of

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two or more atoms bonded together.

Molecules and Chemical Formulas □

Each molecule has a chemical formula. □ The chemical formula indicates: 1. Which atoms are found in the molecule, and 2.

Chapter 7 Notes □ Chemistry;

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### Formulas And Chemical

#### Chapter 7 - Chemical Names &

Formulas. Monatomic ion. Polyatomic ion. Anion. Cation. A single atom with a positive or negative charge resulting from the loss or gain of electrons. A tightly bound group of atoms that behaves as a unit and has a net charge. Any atom or group of atoms with a

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negative charge. Any atom or group of  
atoms with a positive charge.

formulas chapter 7 chemical names

Flashcards and Study ...

Chapter 7 Chemical Formulas and  
Chemical Compounds Chemists use  
chemical names and formulas to



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Formulas And Chemical  
Compounds  
Describe the atomic composition of  
compounds Section 1 Chemical  
Names and Formulas There are  
millions of synthetic and natural  
chemical compounds many of which  
still go by their common name or i.e.  
sodium bicarbonate (baking soda),  
calcium carbonate (limestone), sodium

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chloride (table salt) Common names  
however give you no information about  
chemical composition Systematic  
methods were put in place ...

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Chapter 7: Chemical Formulas and

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## Formulas And Chemical Compounds Section 7-1:

### Chemical Names and Formulas 7-1-1

Explain the significance of a chemical formula. The subscript indicates that there are 8 carbon atoms in the molecule of octane. The subscript indicates that there are 18 hydrogen atoms in the molecule of octane.

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Subscript 2 refers to 2 aluminum...

## Compounds

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include: Test for Fe(II) and Fe(III)

Water of Hydration. Determining the  
Empirical. Formula of Magnesium  
Oxide.

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Chemical Compounds

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Chapter 7.1: Chemical names and formulas. chemical formula.

Hydrocarbons. monatomic ion. name the two types of ions. compound-combination of different atoms in a fixed ratio. compound made of carbon and hydrogen atoms. ions made of single atom. cation, anion.

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chemical formulas chapter 7 1  
Flashcards and Study Sets ...

"A chemical formula indicates the \_\_\_\_\_ number of \_\_\_\_\_ of each kind in a chemical compound.", What do subscripts tell you about a chemical compound?, The polyatomic ion

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chlorite ( $\text{ClO}_2$ )- bonds with copper to form  $\text{Cu}(\text{ClO}_2)_2$ , a compound called copper (II) chlorite. How many atoms of each element are there in copper chlorite?, Write down the name of the following ion:  $\text{V}^{4+}$ , and determine whether ...



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Chapter of the first semester. It serves  
as a summary of all course content  
discussed so far, as

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