

Chapter 12 Chemistry Matter And Change Answers

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Chemistry: Matter and Change

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The two categories that can be used to classify matter are extensive and intensive. An extensive properties depends on the amount of matter, while intensive properties depend of the type or kind of matter, rather than how much of it there is. Work Step by Step The two categories that can be used to classify matter are extensive and intensive.

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Chapter 12 States of Matter Chemistry Matter and Change. Kinetic - molecular theory. Elastic collision. Kinetic energy formula. Temperature. Describes the behavior of matter in terms of particles in moti.... No kinetic energy is lost. KE=1/2mv². Measure of the average kinetic energy of the particles in a sa....

chapter 12 vocabulary chemistry matter Flashcards and ...

1. increasing the surface area. 2. increasing the temperature. 3. increasing the strength of the intermolecular forces. 1 and 2 only. Liquids that have high vapor pressure and low boiling points are called. volatile liquids. The amount of heat required to melt one mole of a solid is called the: heat of fusion.

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