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1. A mixture (is/is not)
a chemical combining
of substances. 2. In a
compound the

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(atoms/molecules) are
(chemically/physically
) combined so that the
elements that make
up the compound
(retain/lose) their
identities and (do/do
not) take on a new set
of properties. 3.

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Supplemental
Problems Answer Key
Chemistry: Matter and
Change 55 b. Ca n 5
4 c. Sn n 5 5 15. Write
the electron
configuration for each

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element described below and identify the element. a. an element that contains 8 electrons $1s^2 2s^2 2p^4$ b. an element that contains 14 electrons $1s^2 2s^2 2p^6 3s^2 3p^2$ The element is silicon.

Chapter 6

~~ANSWER KEY~~
~~HONORS~~

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~~CHEMISTRY~~

(b) Calculate the value of ΔH° , in kJ mol^{-1} , for the combustion of glucose. $2.50 \text{ g} \times 612$
 $6 \text{ mol C}_6\text{H}_{12}\text{O}_6 \text{ } 180.16$
 $\text{g C}_6\text{H}_{12}\text{O}_6 \text{ } 612 \text{ } 6 =$
 $0.0139 \text{ mol C}_6\text{H}_{12}$
 $\text{O}_6 \text{ } 39.0 \text{ kJ } 0.0139$
 $\text{mol}^{-1} = -2,810 \text{ kJ}$
 mol^{-1} 1 point is earned for the correct answer. (c) When

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oxygen is not available, glucose can be oxidized by fermentation. In that process, ethanol and

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Scoring Guidelines
Form B~~

Answer Key 1.c 2.a
3.c 4.b 5.d 6.false
7.macroscopic 8.true
9.false 10.true
11.false 12.applied

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13. pure 14. c 15. a. the initial research was most likely a pure chemistry type of project, learning how to make polymers and studying the properties of these reactions. b. the applied chemistry came in when they wanted to make a specific product with

...

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~~GK-12 Chemistry~~

(a) (b) Answer is one
of the following.

A/yellow: mass =
65.14 kg, volume =
3.38 L, density = 19.3
kg/L, likely identity =
gold. B/blue: mass =
0.64 kg, volume =
1.00 L, density = 0.64
kg/L, likely identity =
apple.

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1 - Chemistry 2e |
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Explain your answer.
Both the reactant and
product contains
COOH group, while
the reactant has -OH
group from alcohol
and the product does
not. Spectrum A has a
broad peak at around
3200 cm^{-1} indicating
the -OH group of

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alcohol. Spectrum B
has a broad peak at
2900 cm^{-1} indicating
the -OH group of acid.

~~[Answer Key]~~

~~Analytical Chemistry
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